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#### Abstract

The isomeric 2-substituted-7(5)-methyl-2,3-dihydro-5(7)H-oxazolo[3,2-a]pyrimidin-5-ones 3a-b and 7 -ones $\mathbf{2 a - b}, 7 \mathbf{a}$ were synthesized by cyclocondensation from the 5 -substituted-2-amino-2-oxazolines $\mathbf{1 a - b}$ with biselectrophiles. In boiling ethanol, the reaction of 1a-b with acetylenic esters led to a mixture of $\mathbf{2 a - b}, 7 \mathbf{a}$ with a small amount of $(E)$-2-N-(2-ethoxycarbonylethylene)-5-substituted-2-iminooxazolines 5a-b. The ring annulation between 1a-b and diketene gave the 2 -substituted-7-hydroxy-7-methyl-2,3,6,7-tetrahydro- 5 H -oxazolo[3,2-a] pyrimidin-5-ones $\mathbf{4 a - b}$ which can be easily dehydrated to provide the 2-substituted-7-methyl-2,3-dihydro-5H-oxazolo[3,2-a]pyrimidin-5-ones 3a-b.


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Interest in the synthesis of nucleoside analogues stems from their antiviral and antitumoral activities [1-4]. Much effort has been devoted to simplify the structures leading to the synthesis of various acyclonucleosides. New pyrimidinones have been defined as synthons to produce new classes of acylnucleosides by reaction with the appropriate nucleophiles [5-7]. We have recently developed the preparation of related bicyclopyrimidinones based on the reactivity of the amidine moiety of 2-amino-2-oxazolines [8-9].

In this work we report the one-step ring-annulation of 5-[(2-methylphenoxy)methyl]- 1a and 5-[(1-phenyl-4-piperazinyl)methyl]-2-amino-2-oxazoline 1b with
potent biselectrophiles, acetylenic esters or diketene, leading to the isomeric 2 -substituted-5-methyl-2,3-dihydro-7(5) H -oxazolo[3,2-a]pyrimidin-7-ones 2a-b,7a and -5 -ones $\mathbf{3 a}-\mathbf{b}$. The raw materials $\mathbf{1 a} \mathbf{- b}$ were prepared according to the method previously described by the authors [10-11].

In refluxing ethanol, the condensation of $\mathbf{1 a - b}$ with ethyl 2-butynoate afforded the corresponding 2 -substituted-5-methyl-2,3-dihydro- 7 H -oxazolo[3,2-a]pyrimidin-7-ones 2a-b in $48 \%$ and $57 \%$ yield respectively, and to the ( $E$ )-2N -(1-methyl-2-ethoxycarbonylethylene)-5-substituted-2iminooxazolines 5a-b in lower yields (10-11\%). The corresponding pyrimidin-5-ones 3a-b were easily obtained


Figure 1. Side view of the crystal structure of $\mathbf{3 a}$ with our numbering scheme, displacement ellipsoids are drawn at the $30 \%$ probability level.

Scheme 1


1a-b, $40-53 \%$




5a-b $\left(\mathrm{R}^{\prime}=\mathrm{CH}_{3}\right), 10-12 \%$
$6 \mathbf{a}\left(\mathrm{R}^{\prime}=\mathrm{H}\right), 10 \%$
3a-b, 22-43\%

a

b

Synthesis of compounds 2a-b and 3a-b.

Table 1
Crystal Data and Structure Refinement for Compound 3a

| Formula | $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{3}$ |
| :---: | :---: |
| Molecular weight | 272.30 |
| Crystal size (mm) | $0.40 \times 0.15 \times 0.10$ |
| Lattice | monoclinic |
| Sp. Gr. | $\mathrm{P} 21 / \mathrm{c}$ |
| a (A) | 12.817(2) |
| b (A) | 14.5120(10) |
| c (A) | 7.487(2) |
| $\mathrm{b}\left({ }^{\circ}\right)$ | 99.86(2) |
| D ( $\mathrm{mg} / \mathrm{m}^{-3}$ ) | 1.318 |
| $\mathrm{F}(000)$ | 576 |
| Z | 4 |
| Temperature | 296(2) K |
| $\mu\left(\mathrm{mm}^{-1}\right)$ | 0.763 |
| h, k, 1 | 0 15, 0 17, -8 +8 |
| 1 (Å) | 1.54178 |
| No. of reflections | 2330 |
| No. of observed reflections | 2330 |
| Volume ( $\AA^{3}$ ) | 1372.0(4) |
| No. of variables | 182 |
| R (F) | 0.0457 |
| Rw (F) | 0.1329 |

by reaction of $\mathbf{1 a - b}$ with diketene via the intermediates 2-substituted-7-hydroxy-7-methyl-2,3,6,7-tetrahydro-5H-oxazolo[3,2-a]pyrimidin-5-ones 4a-b. By performing the reaction at $0^{\circ} \mathrm{C}$ in acetone, we succeded in isolating these unstable cyclic hemiaminals $\mathbf{4 a - b}$, which result from a concerted addition [12-13].

Structural elucidation of 2a-b was achieved by ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C} \mathrm{nmr}$ on the basis of previous results for the 2-aryl-2,3-dihydrooxazole[3,2-a]pyrimidin-7-one series published by Agami et al. [5-7]. The ${ }^{1} \mathrm{H} n m r$ spectrum of 2a showed, at 5.75 ppm , the characteristic singlet for the proton at position 6, whereas it was found at 5.89 ppm for 3a. In the ${ }^{13} \mathrm{C} \mathrm{nmr}$ spectrum, the signals at 107.0 ppm ( $\mathrm{C}-6$ ), $160.8 \mathrm{ppm}(\mathrm{C}=\mathrm{N})$ and $172.5 \mathrm{ppm}(\mathrm{C}=\mathrm{O})$ confirmed the structure of 2a. Different values of the $\mathrm{C}=\mathrm{O}$ were also noticed in the ${ }^{13} \mathrm{C}$ nmr spectra of both compounds $\mathbf{2 a}$ and $\mathbf{3 a}$, at 172.5 ppm for the first one and at 166.1 ppm for the second one.

Some differences were observed in the ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectra of the two regioisomers $\mathbf{2 b}$ and $\mathbf{3 b}$, i.e., the H-6 signal was observed at 5.72 ppm for $\mathbf{2 b}$ and at 5.64 ppm for $\mathbf{3 b}$. The pyrimidin-5-one structure of $\mathbf{3 a}$ has been confirmed by

Table 2
Atomic Coordinates (x $10^{4}$ ) and Equivalent Isotropic Displacement Parameters $\left(\AA^{2} \times 10^{3}\right)$ for Compound 3a

| atom | x | y | z | $\mathrm{U}(\mathrm{eq})$ |
| :--- | :---: | :---: | :---: | :---: |
|  |  |  |  |  |
| $\mathrm{C}(1)$ | $4267(2)$ | $1253(1)$ | $10505(3)$ | $50(1)$ |
| $\mathrm{C}(2)$ | $4777(2)$ | $1092(2)$ | $12267(3)$ | $55(1)$ |
| $\mathrm{C}(3)$ | $5872(2)$ | $1100(2)$ | $12596(4)$ | $73(1)$ |
| $\mathrm{C}(4)$ | $6448(2)$ | $1263(2)$ | $11226(5)$ | $79(1)$ |
| $\mathrm{C}(5)$ | $5930(2)$ | $1440(2)$ | $9519(5)$ | $80(1)$ |
| $\mathrm{C}(6)$ | $4829(2)$ | $1444(2)$ | $9134(4)$ | $70(1)$ |
| $\mathrm{C}(7)$ | $4140(2)$ | $924(2)$ | $13749(3)$ | $81(1)$ |
| $\mathrm{O}(8)$ | $3179(1)$ | $1214(1)$ | $10281(2)$ | $57(1)$ |
| $\mathrm{C}(9)$ | $2572(2)$ | $1360(2)$ | $8539(2)$ | $51(1)$ |
| $\mathrm{C}(10)$ | $1444(2)$ | $1156(1)$ | $8695(2)$ | $43(1)$ |
| $\mathrm{O}(11$ | $1331(1)$ | $158(1)$ | $8899(2)$ | $43(1)$ |
| $\mathrm{C}(12)$ | $1310(1)$ | $-24(1)$ | $10657(2)$ | $33(1)$ |
| $\mathrm{N}(13)$ | $1181(1)$ | $756(1)$ | $11565(2)$ | $35(1)$ |
| $\mathrm{C}(14)$ | $1093(2)$ | $1561(1)$ | $10380(2)$ | $41(1)$ |
| $\mathrm{N}(15)$ | $1383(1)$ | $-848(1)$ | $11284(2)$ | $40(1)$ |
| $\mathrm{C}(16)$ | $1336(1)$ | $-901(1)$ | $13121(2)$ | $40(1)$ |
| $\mathrm{C}(17)$ | $1222(2)$ | $-157(1)$ | $14154(2)$ | $45(1)$ |
| $\mathrm{C}(18)$ | $1131(2)$ | $752(1)$ | $13413(2)$ | $43(1)$ |
| $\mathrm{C}(19)$ | $1432(2)$ | $-1853(2)$ | $13885(3)$ | $60(1)$ |
| $\mathrm{O}(20)$ | $1014(2)$ | $1481(1)$ | $14186(2)$ | $66(1)$ |

Table 3
Bond lengths $(\AA)$ and Angles $\left({ }^{\circ}\right)$ for Compound 3a

| $\mathrm{C}(1)-\mathrm{C}(6)$ | $1.379(3)$ | $\mathrm{C}(10)-\mathrm{C}(14)$ | $1.528(3)$ |
| :--- | :---: | :---: | :---: |
| $\mathrm{C}(1)-\mathrm{O}(8)$ | $1.378(2)$ | $\mathrm{O}(11)-\mathrm{C}(12)$ | $1.347(2)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | $1.389(3)$ | $\mathrm{C}(12)-\mathrm{N}(15)$ | $1.283(2)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | $1.383(3)$ | $\mathrm{C}(12)-\mathrm{N}(13)$ | $1.345(2)$ |
| $\mathrm{C}(2)-\mathrm{C}(7)$ | $1.505(4)$ | $\mathrm{N}(13)-\mathrm{C}(18)$ | $1.396(2)$ |
| $\mathrm{C}(3)-\mathrm{C}(4)$ | $1.383(4)$ | $\mathrm{N}(13)-\mathrm{C}(14)$ | $1.460(2)$ |
| $\mathrm{C}(4)-\mathrm{C}(5)$ | $1.360(4)$ | $\mathrm{N}(15)-\mathrm{C}(16)$ | $1.389(2)$ |
| $\mathrm{C}(5)-\mathrm{C}(6)$ | $1.391(4)$ | $\mathrm{C}(16)-\mathrm{C}(17)$ | $1.352(3)$ |
| $\mathrm{O}(8)-\mathrm{C}(9)$ | $1.415(2)$ | $\mathrm{C}(16)-\mathrm{C}(19)$ | $1.492(3)$ |
| $\mathrm{C}(9)-\mathrm{C}(10)$ | $1.499(3)$ | $\mathrm{C}(17)-\mathrm{C}(18)$ | $1.428(3)$ |
| $\mathrm{C}(10)-\mathrm{O}(11)$ | $1.467(2)$ | $\mathrm{C}(18)-\mathrm{O}(20)$ | $1.226(2)$ |
|  |  |  |  |
| $\mathrm{C}(6)-\mathrm{C}(1)-\mathrm{O}(8)$ | $124.4(2)$ | $\mathrm{N}(15)-\mathrm{C}(12)-\mathrm{N}(13)$ | $127.4(2)$ |
| $\mathrm{C}(6)-\mathrm{C}(1)-\mathrm{C}(2)$ | $121.4(2)$ | $\mathrm{N}(15)-\mathrm{C}(12)-\mathrm{O}(11)$ | $121.8(2)$ |
| $\mathrm{O}(8)-\mathrm{C}(1)-\mathrm{C}(2)$ | $114.2(2)$ | $\mathrm{N}(13)-\mathrm{C}(12)-\mathrm{O}(11)$ | $110.79(14)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | $117.7(2)$ | $\mathrm{C}(12)-\mathrm{N}(13)-\mathrm{C}(18)$ | $121.9(2)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(7)$ | $120.1(2)$ | $\mathrm{C}(12)-\mathrm{N}(13)-\mathrm{C}(14)$ | $111.67(14)$ |
| $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{C}(7)$ | $122.1(2)$ | $\mathrm{C}(18)-\mathrm{N}(13)-\mathrm{C}(14)$ | $126.43(14)$ |
| $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(2)$ | $121.6(3)$ | $\mathrm{N}(13)-\mathrm{C}(14)-\mathrm{C}(10)$ | $101.03(14)$ |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{C}(3)$ | $119.5(2)$ | $\mathrm{C}(12)-\mathrm{N}(15)-\mathrm{C}(16)$ | $113.72(14)$ |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | $120.7(3)$ | $\mathrm{C}(17)-\mathrm{C}(16)-\mathrm{N}(15)$ | $123.3(2)$ |
| $\mathrm{C}(1)-\mathrm{C}(6)-\mathrm{C}(5)$ | $119.1(3)$ | $\mathrm{C}(17)-\mathrm{C}(16)-\mathrm{C}(19)$ | $122.1(2)$ |
| $\mathrm{C}(1)-\mathrm{O}(8)-\mathrm{C}(9)$ | $119.4(2)$ | $\mathrm{N}(15)-\mathrm{C}(16)-\mathrm{C}(19)$ | $114.6(2)$ |
| $\mathrm{O}(8)-\mathrm{C}(9)-\mathrm{C}(10)$ | $106.5(2)$ | $\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{C}(18)$ | $121.8(2)$ |
| $\mathrm{O}(11)-\mathrm{C}(10)-\mathrm{C}(9)$ | $108.5(2)$ | $\mathrm{O}(20)-\mathrm{C}(18)-\mathrm{N}(13)$ | $119.6(2)$ |
| $\mathrm{O}(11)-\mathrm{C}(10)-\mathrm{C}(14)$ | $104.20(14)$ | $\mathrm{O}(20)-\mathrm{C}(18)-\mathrm{C}(17)$ | $128.4(2)$ |
| $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(14)$ | $114.4(2)$ | $\mathrm{N}(13)-\mathrm{C}(18)-\mathrm{C}(17)$ | $112.0(2)$ |
| $\mathrm{C}(12)-\mathrm{O}(11)-\mathrm{C}(10)$ | $108.26(13)$ |  |  |
|  |  |  |  |

X-ray crystallography (Figure 1). Bond lengths and angles (Table 3) do not show surprising features. The 2,3-dihydro- 7 H -oxazolo $[3,2-a]$ pyrimidin- 5 -one moiety is
almost planar, the maximum deviation from planarity is found for $\mathrm{C}(12)$ lying 0.007 (2) $\AA$ from the plane $\mathrm{C}(12)$, $\mathrm{N}(13), \mathrm{N}(15), \mathrm{C}(16), \mathrm{C}(17)$ and $\mathrm{C}(18)$.

During the addition of ethyl butynoate, we obtained a slight part of the non-cyclized esters 5a-b isolated as the (E) conformers. This structural result was established by analyzing the spectrum of the corresponding compound $\mathbf{6 a}$ obtained from 1a with ethyl propiolate. This reaction led to the major pyrimidin-7-one $\mathbf{7 a}$ ( $76 \%$ ) with $\mathbf{6 a}$ in poor yield ( $10 \%$ ). The ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectrum of $\mathbf{6 a}$ exhibited the presence of two typical trans olefinic protons at 8.13 and 5.03 ppm with a coupling constant $J$ of 14 Hz . This conformation could explain the relative stability of $\mathbf{5 a - b}, \mathbf{6}$ a towards an ulterior intramolecular cyclization involving the second nucleophilic nitrogen atom [14]. A comparable behavior of related amidines towards other biselectrophiles was previously described [15-17].

In conclusion, we studied the reactivity of 5 -substituted-2-amino-2-oxazolines 1a-b as synthons to provide new pyrimidinones 2a-b,3a-b,7a. The 2 -substituted-7-methyl-2,3-dihydro-5H-oxazolo[3,2-a]pyrimidin-5-ones 3a-b were easily obtained from diketene after dehydratation of the intermediates $\mathbf{4 a} \mathbf{- b}$. On the other hand, the second regioisomers 2 -substituted-5-methyl-2,3-dihydro-7H-oxa-zolo[3,2-a]pyrimidin-7-ones 2a-b,7a were generated from acetylenic esters by a ring-annulation involving the endocyclic nitrogen atom of the 2 -amino-2-oxazolines 1a-b. During this reaction, the non-cyclized ( $E$ )-2-N-(2-ethoxycarbonylethylene)-5-substituted-2-iminooxazolines $\mathbf{5 a} \mathbf{- b , 6 a}$ were isolated in slight yields. These methodologies are particulary adapted to the synthesis of such heterocycles which may find applications either as bioactive compounds or as useful intermediates for further transformations.

## EXPERIMENTAL

Melting points were determined with a Kofler hot stage apparatus and are uncorrected. The ir spectra were obtained with a Bruker IFS 25 spectrophotometer. The nmr data were recorded with a Bruker AC-200 spectrometer. Chemical shifts ( $\delta$ in ppm) and coupling constants ( J in Hz) were measured using TMS as the internal standard. Silica gel SDS 60 (70-230 mesh) was used for column chromatography. Microanalyses were carried out at the Service central d'analyse CNRS, Vernaison, France.
Crystal Structure Determinations.
Colourless single crystal of 3a was obtained by slow evaporation from methanol solution. The crystallographic data are presented in Table 1. In both cases, the unit cell dimensions were determined using the least-squares fit from 25 reflections ( $\mathrm{q}<25^{\circ}$ ). Intensities were collected with an Enraf-Nonius CAD-4 diffractometer using monochromated $\mathrm{CuK} \alpha$ radiation by the $\mathrm{w} / 2 \mathrm{q}$ scan technique to a limit of $65^{\circ}$. The intensities were corrected for Lorentz and polarization effects, and empirical ( $\Psi$ scans) absorption correction was applied. Both structures were determined by direct methods using MULTAN 80 [18]. The
scattering factors were taken from [19]. C, N and O Atoms were refined anisotropically. The H -atoms were placed in theoritical positions or were located from difference Fourier maps were refined isotropically. The convergence largest $\mathrm{D} / \mathrm{s}$, were $<1$ (on Bs ), the highest peaks in final difference maps were 0.318 and -0.300 e. $\AA^{-3}$. The atomic coordinates presented in Table 2 have been deposited at the Cambridge Crystallographic Data Centre. University Chemical Laboratory, 12 Union Road, Cambridge CB2 IEZ, U.K.

General Procedure for Preparation of 2-Substituted-2,3-dihydro-7H-oxazolo[3,2-a]pyrimidin-7-one 2a-b/7a and ( $E$ )-2-N-(2-Ethoxycarbonylethylene)-5-substituted-2-iminooxazolines 5a-b/6a.

To a solution of 2-amino-2-oxazoline ( $\mathbf{1 a - b}$ ) ( 30 mmol ) in ethanol ( 50 mL ) was added ethyl 2-butynoate or ethyl propiolate $(40 \mathrm{mmol})$. The reaction mixture was heated at reflux for 6 hours. The ethanol was evaporated in vacuo and the residue was separated by column chromatography on silica gel (chloroform/methanol: $90 / 10-\mathrm{v} / \mathrm{v}$ ) to give $\mathbf{2 a - b} / \mathbf{7 a}$ and $\mathbf{5 a - b} / \mathbf{6 a}$.

2-[(2-Methylphenoxy)methyl]-5-methyl-2,3-dihydro-7H-oxazolo[3,2-a]pyrimidin-7-one (2a).

Compound 2a was obtained as colourless crystals (Toluene), yield 48\%; mp 148-149 ${ }^{\circ} \mathrm{C}$; ir (potassium bromide): v 1660 (CO). ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 7.10(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=7.90 \mathrm{~Hz}, \mathrm{H}-5$ '), 7.08 (d, 1H, J = $7.90 \mathrm{~Hz}, \mathrm{H}^{-6}$ '), 6.86 (t, 1H, J = $\left.7.90 \mathrm{~Hz}, \mathrm{H}-4^{\prime}\right)$, 6.75 (d, 1H, J = $7.90 \mathrm{~Hz}, \mathrm{H}-3$ ) , 5.75 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H}-6$ ), 5.28 (m, 1H, $\mathrm{H}-2), 4.43$ ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H}-3$ ), $4.26\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-3\right.$ and $\mathrm{OCH}_{2}$ ), 4.17 (m, $1 \mathrm{H}, \mathrm{OCH}_{2}$ ), $2.19\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.98$ ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}: \delta$ 172.5 (CO), 160.8 (C=N), 155.6 (C-5), 146.7 (C-1'), 130.9 (C-3'), 126.9 (C-5'), 126.5 (C-2'), 121.5 (C-4'), 110.8 (C-6'), 107.0 (C-6), $75.4(\mathrm{C}-2), 67.6\left(\mathrm{OCH}_{2}\right), 46.2(\mathrm{C}-3), 17.6\left(\mathrm{CH}_{3}\right), 15.7\left(\mathrm{CH}_{3}\right)$.
Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{3}$ : C, 66.18; H, 5.88; N, 10.29. Found: C, 66.40; H, 5.95; N, 10.15.

2-(1-Phenyl-4-piperazinylmethyl)-5-methyl-2,3-dihydro-7H-oxazolo[3,2-a]pyrimidin-7-one (2b).

Compound 2b was obtained as colourless crystals, yield $36 \%$; $\mathrm{mp} 93-95{ }^{\circ} \mathrm{C}$; ir (potassium bromide): $v 1660$ (CO). ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 7.23$ (t, 2H, J $=7.60 \mathrm{~Hz}, \mathrm{H}-3^{\prime}$ and $\left.\mathrm{H}-5^{\prime}\right)$, $6.87\left(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=7.60 \mathrm{~Hz}, \mathrm{H}-2^{\prime}\right.$ and $\left.\mathrm{H}-6^{\prime}\right), 6.83(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=7.60 \mathrm{~Hz}, \mathrm{H}-$ $4^{\prime}$ ), 5.72 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H}-6$ ), 5.07 (m, 1H, H-2), $4.30(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=11.55$ and $8.90 \mathrm{~Hz}, \mathrm{H}-3 \mathrm{a}$ ), 4.02 (dd, $1 \mathrm{H}, \mathrm{J}=11.55$ and $7.10 \mathrm{~Hz}, \mathrm{H}-3 \mathrm{~b}), 3.13$ (t, 4H, J = $4.85 \mathrm{~Hz}, \mathrm{CH}_{2}$ pip.), $2.84\left(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=5.25 \mathrm{~Hz}, \mathrm{NCH}_{2}\right.$ ), 2.73 (m, 4H, CH2 pip.), 2.18 (s, 3H, CH3 ); ${ }^{13} \mathrm{C} \mathrm{nmr:} \delta 166.2$ (CO), 161.2 (C=N), 152.0 (C-7), 143.5 (C-1'), 129.1 (C-3' and C-5'), 120.0 (C-4'), 116.2 (C-2' and C-6'), 105.7 (C-6), $77.5(\mathrm{C}-2), 60.3\left(\mathrm{NCH}_{2}\right)$, $54.2\left(\mathrm{CH}_{2}\right.$ pip.), $49.2\left(\mathrm{CH}_{2}\right.$ pip.), $45.8(\mathrm{C}-3), 24.0\left(\mathrm{CH}_{3}\right)$.
Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{2}: \mathrm{C}, 66.18 ; \mathrm{H}, 6.74 ; \mathrm{N}, 17.16$. Found: C, 66.50; H, 6.73; N, 17.28.

2-[(2-Methylphenoxy)methyl]-2,3-dihydro-7H-oxazolo[3,2-a]-pyrimidin-7-one (7a).

Compound 7a was obtained as colourless crystals (Toluene), yield $76 \%$; mp 51-52 ${ }^{\circ} \mathrm{C}$; ir (potassium bromide): v 1650 (CO). ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 7.31$ (d, 1H, J = $7.40 \mathrm{~Hz}, \mathrm{H}-5$ ), 7.05 (m, 2H, H-3' and H-4'), 6.87 ( $\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=7.50 \mathrm{~Hz}, \mathrm{H}-5^{\prime}$ ), 6.71 (d, 1H, J = 7.50 Hz, H-6'), 5.93 (d, 1H, J = 7.40 Hz, H-6), 5.30 (m, 1H, H-2), 4.50-4.11 (m, 4H, 2 $\mathrm{CH}_{2}$ ), $1.94\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.
Anal. Calcd. for $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{3}$ : C, 65.10; $\mathrm{H}, 5.46$; $\mathrm{N}, 10.85$. Found: C, 65.18; H, 5.51; N, 10.78.
(E)-2-N-(1-Methyl-2-ethoxycarbonylethylene)-5-[(2-methylphenoxy)methyl]-2-iminooxazolines (5a).

Compound 5a was obtained as pale yellow oil, yield $10 \%$; ir (potassium bromide): v 3320 (NH), 1765 (CO). ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 7.13(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-5$ ' and H-6'), $6.90(\mathrm{~m}, 1 \mathrm{H}$, H-4'), $6.75\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3^{\prime}\right), 5.61(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}=), 4.79(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH})$, 4.20-3.81 (m, 6H, $3 \mathrm{CH}_{2}$ ), 2.77 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ), 2.18 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ), $1.25\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.0 \mathrm{~Hz}, \mathrm{CH}_{3}\right)$.

Anal. Calcd. for $\mathrm{C}_{17} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{4}$ : C, 64.13; H, 6.96; N, 8.80. Found: C, 64.27; H, 7.05; N, 8.92 .
(E)-2-N-(1-Methyl-2-ethoxycarbonylethylene)-5-(1-phenyl-4-piperazinylmethyl)-2-iminooxazolines ( $\mathbf{5 b}$ ).

Compound 5b was obtained as a yellow oil, yield $11 \%$; ir (potassium bromide): v 3325 (NH), 1720 (CO). ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 7.24\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=7.50 \mathrm{~Hz}, \mathrm{H}-3^{\prime}\right.$ and $\left.\mathrm{H}-5^{\prime}\right)$, $6.89\left(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=7.50 \mathrm{~Hz}, \mathrm{H}-2^{\prime}\right.$ and $\left.\mathrm{H}-6^{\prime}\right), 6.84(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=7.50 \mathrm{~Hz}$, $\mathrm{H}-4 \mathrm{C}), 5.59(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}=), 4.60(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}), 4.25-4.07(\mathrm{~m}, 4 \mathrm{H}$, $2 \mathrm{CH}_{2}$ ), $3.85\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}_{2}\right), 3.59(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=9.30$ and 7.25 Hz , $\mathrm{CH}_{2}$ ), 3.16 (m, 4H, CH2 pip.), 2.71 (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), 2.65 ( $\mathrm{m}, 4 \mathrm{H}$, $\mathrm{CH}_{2}$ pip.), $1.29\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.10 \mathrm{~Hz}, \mathrm{CH}_{3}\right)$.

Anal. Calcd. for $\mathrm{C}_{20} \mathrm{H}_{28} \mathrm{~N}_{4} \mathrm{O}_{3}$ : C, 64.49; H, 7.58; N, 15.04. Found: C, 64.57; H, 7.63; N, 14.91 .
(E)-2-N-(2-Ethoxycarbonylethylene)-5-[(2-methyl-phenoxy)methyl]-2-iminooxazolines (6a).

Compound 6a was obtained as a pale yellow oil, yield $10 \%$; ir (potassium bromide): v 3340 (NH), 1770 (CO). ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 8.14$ (d, 1H, J = $14.0 \mathrm{~Hz}, \mathrm{CH}=$ ), 7.20 (m, 2H, H-3' and H-4'), $6.88\left(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=7.70 \mathrm{~Hz}, \mathrm{H}-5^{\prime}\right), 6.75$ (d, 1H, J = $7.70 \mathrm{~Hz}, \mathrm{H}^{\prime} \mathrm{6}^{\prime}$ ), $5.54(\mathrm{bs}, 1 \mathrm{H}, \mathrm{NH}), 5.03(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $14.0 \mathrm{~Hz}, \mathrm{CH}=), 4.93\left(\mathrm{~m}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right), 3.75\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.14$ ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ), $1.27\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.10 \mathrm{~Hz}, \mathrm{CH}_{3}\right)$.

Anal. Calcd. for $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{4}$ : C, 63.14; H, 6.62; N, 9.20. Found: C, 63.27; H, 6.81; N, 8.97.

General Procedure for Preparation of 2-Substituted-7-methyl-2,3-dihydro-5H-oxazolo[3,2-a]pyrimidin-5-one (3a-b).

A solution of 2-substituted-7-hydroxy-7-methyl-2,3,6,7-tetrahydro- 5 H -oxazolo[3,2-a]pyrimidin-5-one (4a-b) ( 30 mmoles ) in 50 ml of acetone was stirred at $25^{\circ} \mathrm{C}$ during 24 hours. The solvent was removed and the oily residue was triturated in diethyl ether to provide 3a and 3b.

2-[(2-Methylphenoxy)methyl]-7-methyl-2,3-dihydro-5H-oxazolo[3,2-a]pyrimidin-5-one (3a).

Compound 3a was obtained as colourless crystals (Toluene), yield $43 \%$; mp 125-127 ${ }^{\circ} \mathrm{C}$; ir (potassium bromide): v 1685 (CO). ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 7.09$ (t, 1H, J = $7.50 \mathrm{~Hz}, \mathrm{H}-5^{\prime}$ ), 7.07 (d, 1H, J = $7.50 \mathrm{~Hz}, \mathrm{H}-6^{\prime}$ ), 6.85 ( $\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=7.50 \mathrm{~Hz}, \mathrm{H}-4^{\prime}$ ), 6.72 (d, 1H, J = $\left.7.50 \mathrm{~Hz}, \mathrm{H}-3^{\prime}\right), 5.89$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H}-6$ ), $5.24(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2), 4.31\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{H}-3\right.$ and $\left.\mathrm{OCH}_{2}\right)$, $4.13\left(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=10.70\right.$ and $\left.3.05 \mathrm{~Hz}, \mathrm{OCH}_{2}\right), 2.19(\mathrm{~s}, 3 \mathrm{H}$, $\left.\mathrm{CH}_{3}\right), 1.97\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right) ;{ }^{13} \mathrm{C} \mathrm{nmr}: \delta 166.1(\mathrm{CO}), 160.9(\mathrm{C}=\mathrm{N})$, 158.9 (C-7), 155.6 (C-1'), 130.9 (C-3'), 126.8 (C-2' and C-5'), 121.5 (C-4'), 110.7 (C-6'), 105.5 (C-6), 76.2 (C-2), 67.8 $\left(\mathrm{OCH}_{2}\right), 44.1(\mathrm{C}-3), 23.9\left(\mathrm{CH}_{3}\right), 15.6\left(\mathrm{CH}_{3}\right)$.

Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{3}$ : C, 66.18; H, 5.88; $\mathrm{N}, 10.29$. Found: C, 66.39; H, 6.13; N, 10.42.

2-(1-Phenyl-4-piperazinyl)methyl-7-methyl-2,3-dihydro-5H-oxazolo[3,2-a]pyrimidin-5-one (3b).
Compound 3b was obtained as colourless crystals, yield 57\%; mp 130-131 ${ }^{\circ} \mathrm{C}$; ir (potassium bromide): v 1695 (CO). ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 7.26\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=7.65 \mathrm{~Hz}, \mathrm{H}-3^{\prime}\right.$ and $\left.\mathrm{H}-5^{\prime}\right)$, $6.92\left(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=7.65 \mathrm{~Hz}, \mathrm{H}-2^{\prime}\right.$ and $\left.\mathrm{H}-6 '\right), 6.86(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=7.65 \mathrm{~Hz}$, H-4'), $5.64(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-6), 4.68(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2), 3.87(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=$ 11.60 and $9.00 \mathrm{~Hz}, \mathrm{H}-3 \mathrm{a}), 3.62(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=11.60$ and 7.20 Hz , $\mathrm{H}-3 \mathrm{~b}), 3.20\left(\mathrm{t}, 4 \mathrm{H}, \mathrm{J}=4.95 \mathrm{~Hz}, \mathrm{CH}_{2}\right.$ pip.), 2.76 (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), 2.70 (m, 4H, $\mathrm{CH}_{2}$ pip.), 2.65 (d, $\left.2 \mathrm{H}, \mathrm{J}=5.20 \mathrm{~Hz}, \mathrm{NCH}_{2}\right) ;{ }^{13} \mathrm{C}$ nmr: $\delta 168.1(\mathrm{CO}), 155.9(\mathrm{C}=\mathrm{N}), 154.1(\mathrm{C}-5), 151.1(\mathrm{C}-1 '), 129.2$ (C-3' and C-5'), 119.9 (C-4'), 116.2 (C-2' and C-6'), 99.6 (C-6), $72.3(\mathrm{C}-2), 60.9\left(\mathrm{NCH}_{2}\right), 54.0\left(\mathrm{CH}_{2}\right.$ pip. $), 50.9(\mathrm{C}-3), 49.1\left(\mathrm{CH}_{2}\right.$ pip.), $16.5\left(\mathrm{CH}_{3}\right)$.

Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{2}$ : C, 66.18; $\mathrm{H}, 6.74 ; \mathrm{N}$, 17.16. Found: C, 66.32; H, 6.76; N, 17.25.

General Procedure for the Preparation of 2-Substituted-7-hydroxy-7-methyl-2,3,6,7-tetrahydro-5H-oxazolo[3,2-a]pyrim-idin-5-one (4a-b).

A solution of the required 2 -amino-2-oxazoline 1a-b ( 30 mmoles ) in 50 ml of acetone was cooled to $0^{\circ} \mathrm{C}$. Then, diketene ( 40 mmoles) was slowly added. After the addition was complete, the mixture was stirred at $0{ }^{\circ} \mathrm{C}$. A precipitate was observed after 30 minutes. It was isolated by filtration and dried to yield 4a-b.

2-[2-(Methoxyphenoxy)methyl]-7-hydroxy-7-methyl-2,3,6,7-tetrahydro-5H-oxazolo[3,2-a]pyrimidin-5-one (4a) (racemic form).

Compound 4a was obtained as colourless crystals, yield 64\%; mp 139-141 ${ }^{\circ} \mathrm{C}$; ir (potassium bromide): v $3185(\mathrm{OH}), 1710$ (CO). ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 7.11$ (m, 2H, H-5' and H-6'), $6.91(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4 '), 6.75(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3 '), 5.18$ and $5.01(2 \mathrm{~m}$, $1 \mathrm{H}, \mathrm{H}-2), 4.36-4.01\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CH}_{2}\right), 2.85-2.56\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.13$ $\left(\mathrm{m}, 4 \mathrm{H}, \mathrm{CH}_{3}\right.$ and OH$), 1.52\left(2 \mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.

Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{4}: \mathrm{C}, 62.07 ; \mathrm{H}, 6.21 ; \mathrm{N}, 9.65$. Found: C, 62.15; H, 6.15; N, 9.73.

2-(1-Phenyl-4-piperazinyl)methyl-7-hydroxy-7-methyl-2,3,6,7-tetrahydro-5H-oxazolo[3,2-a]pyrimidin-5-one (4b) (racemic form).

Compound 4b was obtained as colourless crystals, yield 65\%; mp 133-134 ${ }^{\circ} \mathrm{C}$; ir (potassium bromide): v $3200(\mathrm{OH}), 1700$ (CO). ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 7.40(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-3$ ' and H-5'), $7.05\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-2^{\prime}\right.$ and H-6'), $7.01\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4{ }^{\prime}\right), 4.83$ ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H}-2$ ), $3.73(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3 \mathrm{a}), 3.47(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3 \mathrm{~b}), 3.17$ $\left(\mathrm{m}, 4 \mathrm{H}, \mathrm{NCH}_{2}\right), 2.65\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{NCH}_{2}\right), 1.80(\mathrm{bs}, 1 \mathrm{H}, \mathrm{OH}), 1.66$
and $1.65\left(2 \mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right) ;{ }^{13} \mathrm{C} \mathrm{nmr}: \delta 167.3$ and $167.2(\mathrm{CO}), 153.6$ and $153.5(\mathrm{C}=\mathrm{N}), 150.9$ and $150.8\left(\mathrm{C}-1^{\prime}\right), 128.9\left(\mathrm{C}-3^{\prime}\right.$ and $\left.\mathrm{C}-5^{\prime}\right)$, 119.6 (C-4'), 115.9 and 115.8 (C-2' and $\left.\mathrm{C}-6^{\prime}\right), 83.5$ and 83.4 $(\mathrm{C}-7), 75.9$ and $75.7(\mathrm{C}-2), 60.1$ and $60.0\left(\mathrm{NCH}_{2}\right), 53.9$ and 53.8 $\left(\mathrm{CH}_{2}\right.$ pip. $), 49.0$ and $48.9\left(\mathrm{CH}_{2}\right.$ pip. $), 44.5$ and $44.4(\mathrm{C}-3), 42.2$ and $42.1(\mathrm{C}-6), 30.4\left(\mathrm{CH}_{3}\right)$.

Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{~N}_{4} \mathrm{O}_{3}$ : C, 62.71; $\mathrm{H}, 6.97 ; \mathrm{N}, 16.26$. Found: C, 62.95; H, 7.13; N, 16.34.

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